## Exercises for Theoretical physics IV

WiSe 2022/23

Sheet 6

25.11.2022

## **Exercise 13** About entropy (I)

Let us consider two sets a and b of paramagnetic ions. The Hamiltonian function of a set is

$$H_i = \mu_B B_i \sum_{j=1}^{N_i} \sigma_j, \ i = a, b;$$
 (1)

where  $\sigma_j = \pm 1$ . The macrostate of each system is characterised by three variables: the energy, the magnetic field and the number of paramagnetic ions. These variables will be denoted hereafter as  $(U_a, B_a, N_a)$  for the system a, and  $(U_b, B_b, N_b)$  for the system b. Both systems are in contact: their magnetic dipoles interact weakly with each other, whereby an energy exchange between both systems is possible. Here  $U_a$  and  $U_b$  can vary, but the total energy  $U = U_a + U_b$ remains constant within a boundary  $\Delta U$ , since the total system is closed. We assume that the system is in thermal equilibrium and is described by a microcanonical distribution. The probability of system a to have the energy  $E_a$  is

$$p(E_a) = \frac{A_a A_b e^{(S_a + S_b)/k}}{\int A_a A_b e^{(S_a + S_b)/k} \mathrm{d}E_a},\tag{2}$$

where k is a constant and

$$S_i(U_i, B_i, N_i) = kN_i \left[ \frac{1}{2} (1+\rho_i) \ln \frac{2}{1+\rho_i} + \frac{1}{2} (1-\rho_i) \ln \frac{2}{1-\rho_i} \right],$$
(3a)

$$A_{i} = \frac{1}{\mu_{B}B_{i}\sqrt{2\pi N_{i}(1-\rho_{i}^{2})}},$$
(3b)

$$\rho_i = \frac{U_i}{N_i \mu_B B_i},\tag{3c}$$

with i = a, b. Consider the extensive quantity  $x = E_a/N_a$ . For given values of  $B_a$ ,  $B_b$ ,  $N_b/N_a$ and  $U/N_a$ , in the limit  $N_a \to \infty$  the dominant contribution to equation (2) is

$$\ln p(x) \sim N_a f(x),\tag{4}$$

where f(x) is an extensive quantity.

a) Determine the function f(x).

(1 point)

b) Show that f(x) is concave.

(2 points)

c) Determine the Taylor expansion of f around the maximum of f(x). Under what conditions is it justified to truncate the Taylor series after the second order?

(2 points)

## **Exercise 14** About entropy (II)

We consider a system of paramagnetic ions with the internal energy

$$U = -N\mu_B B \frac{e^{2\mu_B B/kT} - 1}{e^{2\mu_B B/kT} + 1}.$$
(5)

a) How do the parameter  $\beta$  defined as

$$\beta = \frac{1}{k} \frac{\partial S}{\partial U},\tag{6}$$

the entropy S and the temperature T change, when the internal energy U (from equation 5) changes continuously from the minimum value  $-N\mu_B B$  to the maximum value  $+N\mu_B B$ ?

**Hint:** the temperature  $T = \pm \infty$  should be considered identical, while  $T = 0^+$  belongs to the ground state and  $T = 0^-$  to the highest state.

(2 points)

b) Exploit the fact that entropy is a concave function of energy to show that two types of systems can be distinguished. The first type has an energy spectrum that is only truncated from below and  $\beta$  varies from  $+\infty$  to  $0^+$ . The second type has both an upward and downward curved spectrum and  $\beta$  varies from  $+\infty$  to  $-\infty$ .

(2 points)

c) Consider two systems of both types with initial temperatures  $\beta_a$  and  $\beta_b > \beta_a$ . Show that when both systems are brought into thermal contact, in equilibrium the mean temperature  $\beta$ , with  $\beta_a < \beta < \beta_b$  is reached. Discuss this result for each of the three possible combinations of the different signs of the temperatures. **Hint:** Consider the curves  $S_a(U_a)$  and  $S_b(U_b)$ ; starting from the final equilibrium state, where  $\beta_a = \beta_b = \beta$ , examine the change in temperatures as energy is transferred from one system to the other.

(2 points)